Eradication of *Plasmodium falciparum* from Erythrocytes by Controlled Reactive Oxygen Species via Photodynamic Inactivation Coupled with Photofunctional Nanoparticles

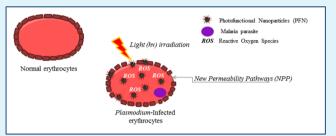
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ABSTRACT: We investigated the antimalarial effect of photodynamic inactivation (PDI) coupled with magnetic nanoparticles (MNPs) as a potential strategy to combat the emergence of drug-resistant malaria and resurgence of malaria after treatment. Because the malarial parasite proliferates within erythrocytes, PDI agents need to be taken up by erythrocytes to eradicate the parasite. We used photofunctional MNPs as the PDI agent because nanosized particles were selectively taken up by Plasmodium-infected erythrocytes and remained within the intracellular space due to the



enhanced permeability and retention effect. Also, the magnetism of Fe_3O_4 nanoparticles can easily be utilized for the collection of photofunctional nanoparticles (PFNs), and the uptaken PFNs infected the erythrocytes after photodynamic treatment with external magnetics. Photofunctionality was provided by a photosensitizer, namely, pheophorbide A, which generates reactive oxygen species (ROS) under irradiation. PAs were covalently bonded to the surface of the MNPs. The morphology and structural characteristics of the MNPs were investigated by scanning electron microscopy and X-ray diffraction (XRD), whereas the photophysical properties of the PFNs were studied with Fourier transform infrared, absorption, and emission spectroscopies. Generation of singlet oxygen, a major ROS, was directly confirmed with time-resolved phosphorescence spectroscopy. To evaluate the ability of PFNs to kill malarial parasites, the PDI effect of PFNs was evaluated within the infected erythrocytes. Furthermore, malarial parasites were completely eradicated from the erythrocytes after PDI treatment using PFNs on the basis of an 8 day erythrocyte culture test.

KEYWORDS: photofunctional nanoparticles, photodynamic inactivation, reactive oxygen species, Plasmodium, malaria, Plasmodium-infected erythrocytes

■ INTRODUCTION

Malaria is a mosquito-borne disease caused by Plasmodium species and is responsible for approximately 800 000 deaths a year worldwide.¹ The malaria epidemic is mediated by Plasmodium-infected mosquitoes; the parasite is transmitted to the human host after a blood meal; in the human host, the malarial protozoans parasitize erythrocytes and reproduce through self-replication.^{2,3}

Various antimalarial drugs have been developed for the treatment of malaria,⁴ such as antifolates, which suppress the synthesis of folic acid, which is essential for the proliferation of Plasmodium.^{1,5} However, these drugs have lost their efficacy due to the emergence of drug-resistant Plasmodium strains induced by overuse of quinolones and antifolates.^{6,7} Several research groups have recently reported the treatment of drug-resistance bacteria using the photodynamic therapy (PDT).^{8–10} In PDT, reactive oxygen species are generated through charge-transfer and/or energy-transfer processes from a photoexcited photosensitizer and induce apoptosis or necrosis of harmful

organisms by oxidation and/or activation of the signaling pathways.^{11–15} Baptista and colleagues showed that PDT could be used to treat various tropical diseases, including malaria.^{8,12} However, because general PDI agents can be taken up by noninfected erythrocytes, this method has limited practical application.¹⁶

In this study, we report enhanced photodynamic inactivation (PDI) of Plasmodium-infected erythrocytes by photofunctional nanoparticles (PFNs) and maintenance of the PDI effect of PFN over an extended period of time.

EXPERIMENTAL METHODS

Fabrication of PFNs. Magnetic nanoparticles (MNPs) were fabricated using the solvothermal reduction method reported previously.¹⁷ FeCl₃· $6H_2O$ (98%), pheophorbide A (PA; 99%),

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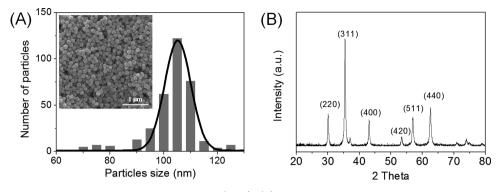


Figure 1. (A) Particle-size distribution of MNPs and SEM image (inset). (B) XRD pattern of MNPs.

polyethylene glycol (PEG; MW 1305–1595), ethylene glycol (EG; 99%), and NaAc (99%) were purchased from Sigma-Aldrich and used without further purification. FeCl₃·6H₂O (1.35 g, 5 mmol) and PEG (1.0 g) were completely dissolved in EG (40 mL), and then, NaAc (4.69 g) was added to the mixture under vigorous stirring (200 rpm) at 60 °C for 30 min. The viscous slurry was transferred into a Teflon bottle in a solvothermal autoclave reactor.^{18,19} The autoclave was heated and maintained at 200 °C for 12 h and then allowed to cool to room temperature. Precipitates were collected after washing with ethanol five times and were dried at 60 °C for 6 h. To fabricate PFNs by the esterification reaction, MNPs (10 mg) were dispersed in the PA solution (10 mL, 1.76×10^{-5} M in tetrahydrofuran (THF)). This solution was agitated (200 rpm) at room temperature. After 24 h, the product was washed with the THF solvent three times and then dried at 60 °C for 6 h.¹⁷

Instrumentation. Field-emission scanning electron microscopy (FE-SEM; JSM-6701F; Jeol Ltd.) was utilized to study the morphology of the MNPs. The crystallographic structure of the nanoparticles was investigated by X-ray diffractometry (XRD; Rigaku, ultima IV) using Cu K α radiation. Fabrication of PFNs was confirmed by Fourier transform infrared (FT-IR) spectroscopy using an FT-IR spectrophotometer (Impact 400; Nicolet). Steady-state absorption and emission spectra were obtained with a UV-vis spectrophotometer (U-2900; Hitachi) and a spectrofluorimeter (F-4500; Hitachi), respectively. The Nd-YAG (surelite II-10, 10 Hz, 7 ns; Continuum)pumped optical parametric oscillator laser (OPO plus, 5 ns; Continuum) was utilized as the excitation source ($\lambda_{ex} = 668$ nm) for the detection of time-resolved singlet oxygen phosphorescence. The phosphorescence signal was collected with a near-infrared photomultiplier tube (H10330A; Hamamatsu) at an angle perpendicular to the excitation beam through cutoff (<1000 nm; CVI) and interference filters (1270 nm; Spectrogon). The signal was amplified and acquired by an amplifier (SR445A; Stanford Research Systems) and a 500 MHz digital oscilloscope (DS07052A; Agilent Technology) and transferred to a computer for further analysis.

Biological Assays. The light source for irradiation of PFNs was a light-emitting diode (LED, $\lambda_{max} = 667$ nm, full width at half-maximum $(FWHM) = 79.6 \text{ nm}, 6.58 \text{ mW/cm}^2$). LED light was radiated onto the PFNs through a cutoff filter (<400 nm; CVI) to block the residual UV light of the LED. The power density of the LED light at the sample position was measured with a power meter (PM200; Thorlabs). To evaluate the PDI efficiency of the PFNs, erythrocytes and Plasmodium falciparum strain 3D7 (chloroquine-susceptible strain) were used. Complete media consisted of Roswell Park Memorial Institute (RPMI) 1640 supplemented with hypoxanthine, NaHCO₃, human plasma, gentamicin, and AlbuMAX I. To infect the erythrocytes, erythrocytes (45.5 \times 10⁶ cell/mL) were cultured with *P. falciparum* 3D7 in complete media under a mixed gas phase $(O_2/CO_2/N_2 =$ 5:5:90) at 37 °C for 3 weeks.²² Infection of erythrocytes was evaluated by Giemsa staining.²³ Of the erythrocytes, 10% were infected with P. falciparum 3D7. The Plasmodium-infected erythrocytes were cultured in the dark with various concentrations of PFNs (0–100 μ g/mL) for toxicity testing and were also cultured without PFNs under various light-irradiation conditions (LED, $\lambda_{max} = 667$ nm, FWHM = 79.6 nm,

 6.58 mW/cm^2 , 0-120 min) to test for light toxicity. The PFNs were incubated with infected erythrocytes at room temperature for 24 h in the dark to allow uptake. To evaluate the intracellular uptake efficiency of PA and PFNs, PA (0.16 μ M) and PFNs (25 μ g/mL) were incubated with Plasmodium-infected erythrocytes for 60 min at 37 °C. The presence of PA and PFNs in the Plasmodium-infected erythrocytes was assessed using a fluorescence microscope (iXon EMCCD; Andor) and flow cytometric analysis (NAVIOS; Beckman Coulter). In the case of flow cytometric analysis, RBCs were gated on the basis of their forward scatter and side scatter signals with logarithmic scales. At least 50 000 particles were assessed and plotted in two-dimensional scattergrams of two of these three parameters, FSC, SSC, and FL4 (Fluorescence Detector 4 (680-710 nm emission)). PA and PFNs were detected in RBC areas with fluorescence (FL2). The RBC area, PA area, and PFNs were detected by analyzing the scattergrams from the computer software.

PDI and Eradication of Plasmodium from Erythrocytes. PDI experiments were performed with various concentrations of PA and PFN. The amount of PA molecules bonded to the surface of the magnetic particles was estimated using UV-vis absorption spectroscopy. Absorption optical densities (O.D.'s) at 667 nm were measured with the initial PA/THF solution $(1.85 \times 10^{-5} \text{ M})$ before the addition of magnetic particles for reaction and also with the PA/THF solution that remained after reaction with the particles. The O.D. difference between the initial and remaining PA/THF solutions indicated that ~6.49 \times 10⁻⁸ mol (equivalent to 3.9 \times 10¹⁶ molecules) of PA was immobilized onto the surface of 10 mg of magnetic particles. Therefore, the number of PA molecules bonded to the surface of the magnetic particles at 25, 50, 75, and 100 μ g/mL of PFN was equivalent to the number of PA molecules in the liquid solution at concentrations of 0.16, 0.32, 0.48, and 0.65 µM, respectively. In experiments comparing PA and PFNs, the number of PA molecules applied was kept the same. The applied concentrations were 0-0.65 μ M PA and 0–100 μ g/mL PFNs. Plasmodium-infected erythrocytes that were treated with PA or PFN were cultured for 24 h and then placed in culture flasks under light irradiation with an LED ($\lambda_{\rm max}$ = 667 nm, FWHM = 79.6 nm, 6.58 mW/cm²) for 30 min. After 48 h, the number of infected cells was estimated by Giemsa staining $^{\rm 23}$ and the Plasmodium real-time polymerase chain reaction (RT-PCR) assay. Eradication of the parasite was validated by evaluating Plasmodium proliferation after 8 days of culture of the Plasmodium-infected erythrocytes treated with PDI. The target genes of the RT-PCR were circumsporozoite protein (CSP) and plasmodium lactate dehydrogenase (pLDH) genes of P. falciparum. Plasmodium DNAs were extracted from 100 μ L of the PDI-treated samples using the QIAamp DNA Mini Kit (Qiagen, Hilden, Germany). Briefly, each tube contained a 25 μ L reaction mix, which included 2.5 μL of isolated DNA, 0.1 μM of forward and reverse primers, and 0.1 μ M of the probe. TaqMan amplification and detection were performed with a real-time thermocycler CFX96 (Bio-Rad, Hercules, CA). Thermocycling conditions were initial denaturation at 95 °C for 3 min, followed by 45 cycles at 95 °C for 15 s and 60 °C for 60 s. The RT-PCR analysis indicates that the lower the threshold cycle (C_t) value, the more the number of genes present.

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RESULTS AND DISCUSSION

As shown in Figure 1A, the fabricated MNPs had good size uniformity. The size histogram of MNPs was obtained by

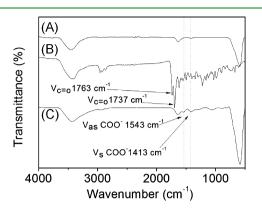


Figure 2. FT-IR spectra of (A) MNPs, (B) PA molecules, and (C) PFNs.

sampling 340 particles in different regions of the FE-SEM image. The average particle size was 103 ± 21 nm. The powder XRD pattern of MNPs provided detailed structural information, as shown in Figure 1B. The strong Bragg reflection peaks ($2\theta = 30.0, 35.6, 43.3, 53.7, 57.0, 62.8^{\circ}$) are marked by their Miller indices ((220), (311), (400), (422), (511), and (440), respectively), which were obtained from standard Fe₃O₄ powder diffraction data (card 19-0629; JCPDS). The position and relative intensity of all diffraction peaks corresponded to the characteristic peaks of a magnetite crystal with a cubic inverse spinel structure.^{17,19}

To understand the bonding nature of the carboxyl groups of PA and the Fe ions of MNPs, FT-IR spectra of all samples were compared, as shown in Figure 2. The IR spectrum of PA had absorption peaks at 1763 and 1737 cm⁻¹, which correspond to the stretching modes of the free carbonyl double bond ($v_{C=O}$) (Figure 2B).²⁴ After esterification between the carboxyl group of PA and Fe ion of MNPs, the peaks assigned to the free carboxyl group disappeared (1763 and 1737 cm⁻¹) and new bands appeared at 1543 and 1413 cm⁻¹, which we ascribed to the asymmetric (v_{as}) and symmetric (v_s) stretch vibrations of the carboxylate group (COO—), respectively.²⁴ The appearance of these new bands was also reported in the IR study of Dravid and colleagues.²⁵ This pattern of spectral change is due to Fe–carboxylate complexation as a result of chemical coordination, as shown previously.²⁶

The photophysical properties of PFNs were characterized by steady-state absorption and emission spectroscopies. Figure 3A presents the characteristics of PA absorption bands: the Soret band at 410 nm and the Q bands at 504, 534, 608, and 666 nm were nearly identical for PA in THF solution and PFNs in THF solution.²⁷ The fluorescence emission peaks of PFNs in THF solution at 675 and 725 nm were also similar to those of PA in THF solution (Figure 3B). This suggests that the photophysical properties of PA are not altered by bonding between PA and MNPs.¹⁰

The most direct measurement method for singlet oxygen is the detection of phosphorescence from the deactivation of singlet oxygen molecules induced by photoexcited PA.²⁸ As shown in Figure 4, singlet oxygen phosphorescence signals from low concentrations of PA (0.016 μ M) and PFNs (2.5 μ g/ mL) were measured in THF solution at a detection wavelength of 1270 nm to avoid light scattering from the MNPs. The intensities of the singlet oxygen phosphorescence signal at the zero time point were nearly identical for PA and PFNs in THF solution. The phosphorescence decay signals from PA and PFNs were fitted to a single exponential function, yielding similar singlet oxygen lifetimes of 22.7 and 22.1 μ s, respectively. These measured singlet oxygen lifetimes are consistent with the reported value.²⁸ These results indicate that the same amount of singlet oxygen was generated from PA and PFN and that the singlet oxygen generated from PFNs did not undergo selfquenching by the interface modified MNPs.^{20,29}

To assess the toxicities of PFNs and irradiated light, Plasmodium-infected erythrocytes were cultured with PFNs in the dark or only with light irradiation in the absence of PFNs. PFNs were not toxic in the concentration range of 0-100 μ g/mL, and no light toxicity in the energy range of 0-48 J was evident (see Figure 5).

The PA- and PFN-uptaken RBC counts are quantified as 11.7 and 23.8%, respectively. The more efficient uptake of PFNs compared to that of PA shown in Figure 6B is expected to be due to the enhanced permeability and retention (EPR) effect of the former.³⁰ The permeability of the erythrocytes increases during malarial infection due to the destruction of the spectrin network.³¹ Both PFNs and PAs can be taken up easily by Plasmodium-infected erythrocytes due to the induction of new permeability pathways by the malarial parasite.³² The uptaken PFNs accumulate in the Plasmodium-infected erythrocytes because of physicochemical pharmacological factors.^{16,32}

After 30 min of irradiation for PDI treatment, the infection rate of the erythrocytes decreased in a dose-dependent manner according to the concentrations of PAs and PFNs (Figure 7A).

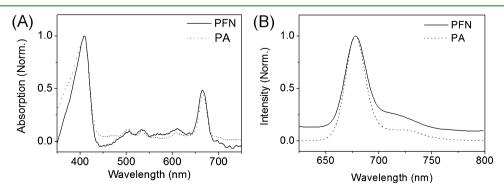


Figure 3. (A) Absorption and (B) emission spectra of pure PA and PFNs in THF. The excitation wavelength for the emission spectrum was 410 nm.

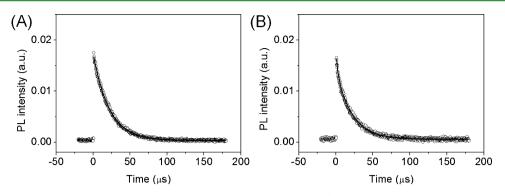


Figure 4. Phosphorescence decay signals induced by relaxation of the singlet oxygen from (A) PA and (B) PFNs in THF solution. Phosphorescence signals were detected at 1270 nm and fitted with a single exponential function (solid line).

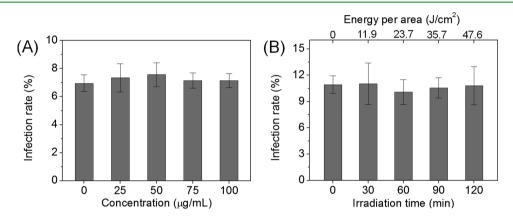


Figure 5. Toxicity test of (A) PFNs without light irradiation and (B) only under light irradiation without PFNs. These experiments were repeated three times.

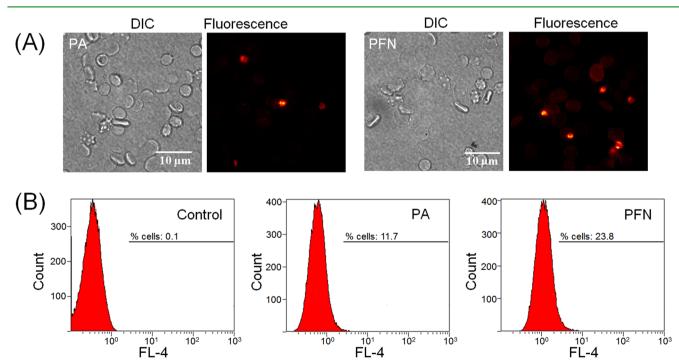


Figure 6. Fluorescence images of PA and PFNs in Plasmodium-infected erythrocytes. (A) Plasmodium-infected erythrocytes were incubated with PA (0.16 μ M) or PFNs (25 μ g/mL) in a culture medium for 1 h at 37 °C. The uptake efficiencies of PA and PFNs were quantified by flow cytometric analysis (B).

As shown in Figure 7B, there was a difference in the reduction of infection rates between PAs (0.32 μ M) and PFNs (50 μ g/mL) after irradiation with LED light. The concentration of PA

 $(0.32 \ \mu\text{M})$ was equivalent to the concentration of PA molecules bonded to the surfaces of the PFNs (50 μ g/mL). PFNs resulted in a greater reduction in infection during the initial 15 min and

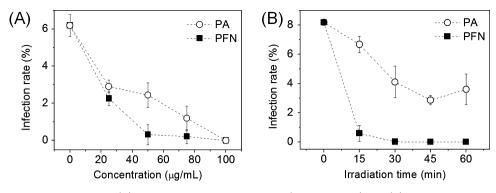


Figure 7. PDI of *P. falciparum* according to (A) the concentration of PDI agents (PAs and PFNs) and (B) LED irradiation time. The experiments were repeated in triplicate.

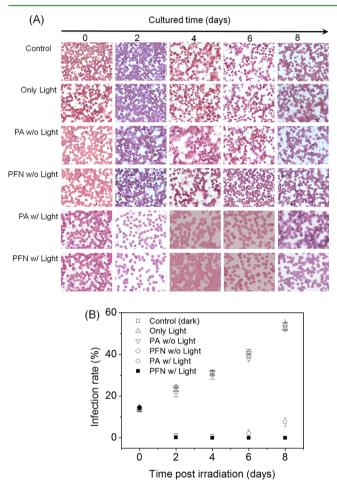


Figure 8. *P. falciparum* eradication at 8 days after PDI treatment. Plasmodium-infected erythrocytes were visualized by Giemsa staining. Magnified stained images of control (dark), only light, PA/PFN without light, and PA-/PFN-mediated PDI treatments are shown in (A). (B) Infection rates were quantified as a function of time. The experiments were repeated in triplicate.

a 0% infection rate after 30 min. In contrast, PA molecules reduced the infection rate to only 56% of that of PFNs over the same time period. This suggests that PFNs are more effective than PAs for PDI treatment of Plasmodium-infected erythrocytes.

The Plasmodium-infected erythrocytes were incubated for 8 more days after only light irradiation for 30 min without PDI agents (PAs or PFNs) and incubated for 8 days with only PA (0.48 μ M) or PFNs (75 μ g/mL) without light irradiation, and

Table 1. Change of the C_t Values of the *P. falciparum* CSP Gene and pLDH Gene by RT-PCR^{*a*}

gene	treatment	0 days	2 days	4 days	6 days	8 days
P.f. ^b CSP ^c gene	control	27.33	29.61	29.62	28.42	24.88
P.f. <i>pLDH^d</i> gene	PA	27.13	33.39	33.8	31.67	29.41
	PFNs	28.5	N/A	N/A	N/A	N/A
	control	27.51	30.12	30.24	28.87	25.06
	PA	27.67	34.04	34.35	32.39	30.09
	PFNs	27.52	N/A	N/A	N/A	N/A

^{*a*}Control, PAs, and PFNs represent the light-only condition without PDI agents and the conditions of PDI treatments with PA and PFNs, respectively. ^{*b*}P. *falciparum*. ^{*c*}Circumsporozoite protein. ^{*d*}Plasmodium lactate dehydrogenase.

also incubated with PDI treatment for only 30 min of the initial period with PAs (0.48 μ M) or PFNs (75 μ g/mL). Every 2 days, the proliferation of the malarial parasites was assessed by Giemsa staining and the RT-PCR assay. As shown in Figure 8B, the infection rates of *P. falciparum* were continuously increased in the cases of the light-only condition without the PDI agents and the PDI agents-only condition without irradiation. In the case of PDI treatment with PA, the infection rate increased again after 4 days and the malarial parasite count increased again after 6 days. In contrast, the infection rate did not increase after PDI treatment using PFNs and erythrocyte rupture did not occur (see Figure 8A). In addition, the RT-PCR analysis indicates that the lower the C_t value, the more the number of genes present, as shown in Table 1.

We fabricated MNPs coupled to PA molecules using a simple modification process. The fabricated PFNs were characterized by SEM, XRD, FT-IR, absorption spectroscopy, and emission spectroscopy. The PFNs generated singlet oxygens, as confirmed by time-resolved near-IR spectroscopy. A comparison of PDI treatment with PAs and PFNs showed that Plasmodium-infected erythrocytes were more effectively treated with PFN/PDI than with PA/PDI due to the EPR effect of PFNs. More importantly, PDI treatment with PFNs eradicated the malarial parasite from the erythrocytes without recurrence after 8 days.

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Author Contributions

All authors contributed to the writing of this manuscript and have approved the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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